# RESPIRATORY PROPERTIES OF BLOOD FROM VOLUNTARILY AND FORCIBLY SUBMERGED XENOPUS LAEVIS

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## Accepted 3 October 1985

### SUMMARY

The respiratory properties of blood from voluntarily diving *Xenopus* seem well matched to the animal's habit of ventilating its lungs in an intermittent fashion. Compared with more terrestrial anurans, the high oxyhaemoglobin affinity ( $P_{50} = 29.6$  Torr, pH 7.73, 25 °C), Bohr effect ( $\Delta log P_{50}/\Delta pH = -0.37$ ) and Haldane effect ( $0.37 \text{ mol } CO_2 \text{ mol}^{-1} O_2$ ) can be viewed collectively as adaptations towards effective blood gas storage during periods of apnoea and blood gas exchange during episodes of air breathing. It appears, therefore, that these biochemical adaptations are linked to the changes in respiratory blood flow that are made possible by the partially divided double circulation of *Xenopus*.

In comparison with blood from voluntarily diving Xenopus, that taken from animals at the end of a 30-min enforced dive was haemoconcentrated and contained 4 and 8 mmol  $l^{-1}$  higher concentrations of true plasma lactate and metabolic acid equivalents respectively. The pH-induced effects of the latter led to reductions in oxyhaemoglobin affinity and blood CO<sub>2</sub> carriage, both of which persisted for up to 4 h following emergence from an enforced dive. The associated haemoconcentration led to a secondary series of effects of which the most obvious were an elevated blood oxygen-carrying capacity and an increased non-bicarbonate buffer slope for true plasma. Marked changes, such as these, were never observed in blood samples taken at various stages of voluntary dives lasting upwards of 30 min.

### INTRODUCTION

Xenopus laevis is an aquatic anuran which relies on periodic episodes of air breathing for the major portion of its oxygen uptake. This behaviour causes substantial oscillations in respiratory gases in lungs, blood and tissues (Shelton & Boutilier, 1982). It has been argued that oscillations of this type are an inevitable consequence of the evolution of air-breathing in fishes and early tetrapods (Shelton, Jones & Milsom, 1985). It also seems likely that the early evolution of a partially

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Key words: *Xenopus*, aerobic dives, anaerobic dives, blood respiratory properties, haemoconcentration.

divided, double circulation is linked to the use of the blood as a store for respiratory gases (Shelton, 1985). If these views are to be sustained, it is important to understand the effects of intermittent ventilation on all parts of the gas exchange and storage systems.

In Xenopus, short (10 min) voluntary dives cause blood  $O_2$  partial pressures to fall from approximately 80 to 40 Torr (Emilio & Shelton, 1974). At times, however, the animal will remain voluntarily submerged for periods of up to 1 h (Boutilier, 1984), though the corresponding details of blood gases are not known. In a recent study in which Xenopus was allowed to dive voluntarily but then restricted from reaching the surface, Emilio & Shelton (1980) found that a mild respiratory acidosis developed over the first 10 min. Longer dives of this sort, up to 30 min duration, caused a combined respiratory and metabolic acidosis, the latter persisting for some hours after the animal surfaced and breathed. Associated with this acidosis was a considerable haemoconcentration resulting from a loss of plasma water to some extravascular compartment. Similar observations have been made on other forcibly submerged amphibians (Jones, 1972; Jones & Mustafa, 1973; Emilio, 1974; Lillo, 1978). These and other studies have led to a view that anaerobic pathways are an integral part of these animals' normal diving metabolism. However, there is an increasing amount of evidence to show that periods of voluntary submergence end before O2 stores are fully depleted (Lenfant & Johansen, 1967; Toews, 1971; Toews, Shelton & Randall, 1971; Emilio & Shelton, 1974, 1980). In the present study, we examine the effects of forced and voluntary dives on the oxygen-carrying and acidbase properties of Xenopus blood in order to understand the changes that can be produced in this important gas transport and storage medium as the animal breathes intermittently.

#### MATERIALS AND METHODS

### Preparation of experimental animals

Male and female specimens of the South African clawed toad, *Xenopus laevis*, were either obtained commercially or selected from a laboratory reared stock. All animals were kept in large communal tanks (20–25°C) and fed on chopped liver twice weekly. At least 1 week prior to experimental procedures, the animals were transferred to laboratory tanks ( $25 \pm 2^{\circ}$ C) where feeding was suspended.

After anaesthesia, induced by immersion in a 0.06 % solution of tricaine methanesulphonate (MS-222, Sandoz; titrated to pH7.0 with NaOH), a catheter was chronically implanted in the femoral artery of each animal (Boutilier, 1984). The toads were then artificially ventilated until they had recovered and they were given at least 24 h in which to acclimate to the experimental chambers ( $25 \pm 0.5$  °C). The P<sub>O<sub>2</sub></sub> of the water was >140 Torr at all stages of the experiments, P<sub>CO<sub>2</sub></sub> < 0.5 Torr (Radiometer electrodes and meters).

### Oxygen and carbon dioxide properties of blood and plasma

Acid-base and oxygenation characteristics of whole blood and plasma were studied by *in vitro* tonometry at 25°C. Individual blood samples were drawn from the femoral cannulae of several animals, pooled to the required volumes, and stored at 4°C for less than 1 h before use. Aliquots having the same haematocrit as the parent blood pool were transferred to tonometers (50 ml round bottom glass flasks) and equilibrated with a wide range of gas mixtures produced by Wösthoff pumps (Bochum, FRG). Blood samples for analyses were drawn after a predetermined 30-min equilibration time, into positive displacement gas-tight Hamilton syringes.

Whole blood oxygen equilibrium curves were constructed at several  $P_{CO_2}$  (and thus pH) levels using the mixing method (Haab, Piiper & Rahn, 1960) as detailed further by Scheid & Meyer (1978). Blood samples were equilibrated against gas mixtures containing various concentrations of CO<sub>2</sub> in air and CO<sub>2</sub> in nitrogen. Samples of known O<sub>2</sub> saturation were analysed for  $P_{O_2}$ ,  $P_{CO_2}$ , pH and haematocrit, with replicate samples from the 2% CO<sub>2</sub>/air-equilibrated tonometer being used for measures of total blood O<sub>2</sub>-carrying capacity (i.e. at approximately 14.5 Torr  $P_{CO_2}$  and 155 Torr  $P_{O_2}$ ). The oxygen content of separated plasma was determined also on samples equilibrated with gas mixtures containing 2% CO<sub>2</sub>-balance air. The relationship between blood O<sub>2</sub>-carrying capacity and haematocrit was assessed over a wide range on blood whose haematocrit was experimentally altered to high and low levels by centrifugation and resuspension at various red blood cell to plasma volume ratios.

In vitro acid-base relationships were determined using the same apparatus as above. Measurements of pH,  $P_{CO_2}$ , lactate concentrations, total CO<sub>2</sub> concentrations and haematocrit were made on blood equilibrated against humidified gas mixtures containing various proportions of CO<sub>2</sub> in either air or nitrogen. The true plasma of each blood sample, obtained by centrifugation of blood in a sealed capillary tube, was analysed for total CO<sub>2</sub> concentration. To allow comparison, the above procedures were carried out on blood obtained from unrestrained, freely-diving animals (67-101 g, N = 31) and from animals at the 30 min mark of an enforced dive (62-93 g, N = 19).

### Blood respiratory properties before, during and following enforced dives

Twelve toads (80-120 g) were prepared with arterial catheters and allowed to recover as before. Blood samples were drawn initially from each animal as it was freely diving and surfacing in air-saturated water. By lowering a Perspex lid over the water surface, the animal was forced to remain submerged for 30 min at which time a second blood sample was taken. The lid was then removed from the surface and additional blood samples  $(400 \,\mu\text{l})$  were drawn at +1, +2 and +4 h following the 'dive'. The determinations were carried out on groups of three animals; the blood samples  $(400 \,\mu\text{l} \,\text{animal}^{-1} \,\text{stage}^{-1})$  were pooled, stored on ice, and within 1 h the blood pool was divided equally between an oxygenated and deoxygenated tonometer. Following equilibration with 2% CO<sub>2</sub>/air or 2% CO<sub>2</sub>/N<sub>2</sub> gas mixtures, known blood O<sub>2</sub> saturations were prepared by the mixing method (approximately 10, 50 and 90% saturations). Measurements made on each of these prepared samples included P<sub>O2</sub>, pH, haematocrit and true plasma concentrations of both total CO<sub>2</sub> and lactate.

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### Analytical procedures

Methodologies for the measurement of haematocrit, pH,  $P_{CO_2}$ , total CO<sub>2</sub>, PO<sub>2</sub> and lactate have been detailed previously (Boutilier, Randall, Shelton & Toews, 1977; Boutilier, McDonald & Toews, 1980; Boutilier & Toews, 1981; Boutilier, 1984). Oxygen contents were determined with a Lex-O<sub>2</sub>-Con analyser (Lexington Instruments, MA) using, for whole blood, a sample volume of 20  $\mu$ l. For low O<sub>2</sub> contents (i.e. plasma), a calibration curve between 0 and 2 vols % was constructed by injecting known volumes of O<sub>2</sub>/N<sub>2</sub> equilibrated distilled water.

Bicarbonate concentrations ([HCO<sub>3</sub><sup>-</sup>]) in true plasma were estimated from our measured values of total CO<sub>2</sub> concentration ([CO<sub>2</sub>]) with the equation [HCO<sub>3</sub><sup>-</sup>]= [CO<sub>2</sub>]-( $\alpha$ CO<sub>2</sub>×P<sub>CO<sub>2</sub></sub>), where  $\alpha$ CO<sub>2</sub> is the plasma carbon dioxide solubility (in mmoll<sup>-1</sup> Torr<sup>-1</sup>) at 25 °C (Reeves, 1976). The proportion of the total blood oxygen-carrying capacity which could be attributed to oxyhaemoglobin binding was obtained by subtracting the dissolved fraction of O<sub>2</sub> in plasma from the blood O<sub>2</sub> capacity (note that all O<sub>2</sub> capacity measurements were made at approximately 14.5 Torr P<sub>CO<sub>2</sub></sub> and 155 Torr P<sub>O<sub>2</sub></sub>).

#### RESULTS

The effective increase in haemoglobin concentration of blood from forcibly submerged *Xenopus laevis* was reflected *in vitro* by large differences in oxygen and carbon dioxide combining properties when compared with blood samples from unrestrained animals (Table 1).

Table 1. In vitro oxygenation and acid-base characteristics of blood samples obtained from voluntarily diving Xenopus and for those drawn at the end of a 30-min enforced dive

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Voluntary	Enforced
$28.3 \pm 0.7$ (25)	$46.2 \pm 2.8$ (21)
$-15.0 \pm 2.1$ (7)	$-24.6 \pm 3.0$ (7)
$9.38 \pm 0.6$ (25)	$15.4 \pm 2.1$ (21)
$0.58 \pm 0.03$ (8)	_
$29.6 \pm 1.3$ (10)	$37.0 \pm 1.9$ (5)
$-0.37 \pm 0.03$ (12)	$-0.29 \pm 0.02$ (7)
$0.95 \pm 0.16$ (11)	5·29 ± 0·54 (9)
	Voluntary $28 \cdot 3 \pm 0.7$ (25) $-15 \cdot 0 \pm 2 \cdot 1$ (7) $9 \cdot 38 \pm 0.6$ (25) $0 \cdot 58 \pm 0.03$ (8) $29 \cdot 6 \pm 1.3$ (10) $-0.37 \pm 0.03$ (12)

Values are means ±1 S.E.M. (no. of determinations on independent blood pools). Temperature = 25 °C. • At 2 % CO<sub>2</sub>-balance air equilibration gas. † At ~14.5 Torr P<sub>CO2</sub>. ‡ At 50 % blood O<sub>2</sub> saturation.

### Blood oxygen characteristics in vitro

Oxygen equilibrium curves for whole blood samples drawn from animals engaged in voluntary diving-emergence behaviour and for those following a 30-min enforced dive are illustrated in Fig. 1. Individual curves contributing to each of the averaged curves shown were determined at six  $O_2$  saturation levels (approximately 10, 20, 50, 60, 70 and 90%). For any given level of  $P_{CO_2}$ , the  $P_{50}$  value of blood from forcibly submerged *Xenopus* was greater than that determined for the freely diving animal, as were the concentrations of lactate and free hydrogen ions (Table 1; Fig. 1). The large haematocrit increase following a forced dive and the corresponding increase in blood  $O_2$ -carrying capacity, meant that this blood *in vitro* carried greater amounts of chemically bound oxygen for any given level of  $P_{O_2}$ .

The CO<sub>2</sub> Bohr effect (Fig. 2) was determined by linear regression analysis of all data points obtained at 50% saturation from the oxygen equilibrium curves of both voluntarily and forcibly submerged animals (Fig. 1). Though the slope of the  $\Delta \log P_{50}/\Delta pH$  relationship was evidently reduced in blood taken from involuntarily diving toads (Fig. 2), statistical comparisons of the slopes are not offered as they represent data collected over two different pH ranges.

The relationship between whole blood  $O_2$ -carrying capacity ( $C_bO_2$ ) and haematocrit (Fig. 3) was described by the equation;  $C_bO_2 = 0.339$  (haematocrit)+0.61 (correlation coefficient, r = 0.985, N = 45). As expected, the y-intercept of this equation is virtually identical to the measurement of the  $O_2$ -carrying capacity of plasma (Table 1; Fig. 3). Data in Fig. 3 are values for freely diving and forcibly submerged animals with additional points at high and low haematocrit values being obtained by experimentally preparing various red blood cell to plasma volume ratios. Note that the points for each group of animals are clustered on discrete portions of the line, the higher values being attributable to the haemoconcentration which occurred during a 30-min enforced dive (Table 1).

### In vitro CO<sub>2</sub> relationships

An increase in  $P_{CO_2}$  causes the red blood cells to swell in *Xenopus* as in mammals. This swelling is reflected as an increased haematocrit as Fig. 4 shows. Determinations of oxyhaemoglobin-carrying capacity made at the same time as the haematocrit showed that it did not change with equilibration gas  $P_{CO_2}$ , confirming that the haematocrit variations were due to cell swelling. Plasma trapping during centrifugation can amount to 2–3% in packed red cells of *Xenopus* (Emilio & Shelton, 1980) and, though this may change as a result of increased red cell size, the effects will be much smaller than those measured here.

Xenopus whole blood exhibited a Haldane effect: at any given  $P_{CO_2}$  level, the whole blood total CO<sub>2</sub> content and pH were significantly lower in oxygenated than deoxygenated blood (Fig. 5). The mean difference in total CO<sub>2</sub> content between oxygenated and deoxygenated blood was found to be  $1.4 \text{ mmol} 1^{-1}$  (0.045-0.070 pHunits difference). Accordingly, the Haldane effect is expressed as the ratio of the change in blood total CO<sub>2</sub> content between oxygenated and deoxygenated blood over

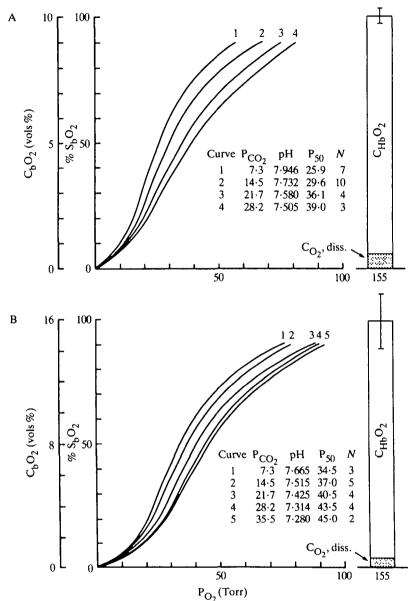


Fig. 1. In vitro oxygen equilibrium curves of whole blood taken from (A) freely diving and (B) forcibly submerged Xenopus at 25°C. Curve numbers are percentages of equilibration gas CO<sub>2</sub>. Tabulated data are mean values for blood acid-base status at the half saturation value P<sub>50</sub>. N, number of independent curve determinations on separate blood pools; C<sub>b</sub>O<sub>2</sub>, whole blood oxygen-carrying capacity at  $\approx$ 155 Torr P<sub>O2</sub>; S<sub>b</sub>O<sub>2</sub>, whole blood O<sub>2</sub> saturation; C<sub>Hb</sub>O<sub>2</sub>, oxyhaemoglobin carrying capacity at  $\approx$ 155 Torr P<sub>O2</sub>; C<sub>O2</sub>, diss., plasma O<sub>2</sub> capacity at  $\approx$ 155 Torr P<sub>O2</sub>. Histogram bars show the mean blood O<sub>2</sub>carrying capacity ( $\pm$ 1 s.E.M.) partitioned into oxyhaemoglobin carrying capacity and plasma O<sub>2</sub> capacity.

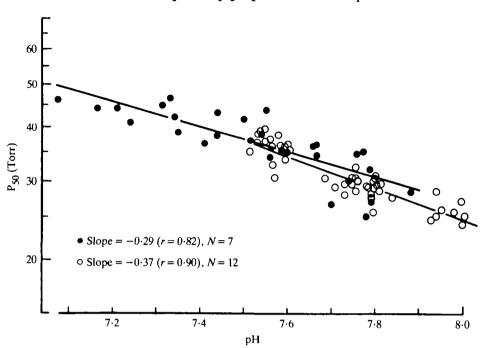


Fig. 2. In vitro CO<sub>2</sub> Bohr effect of whole blood obtained from freely diving (open circles) and forcibly submerged (closed circles) Xenopus at 25°C. Slopes and correlation coefficient, r, determined by least squares regression of the relationship  $\Delta \log P_{50}/\Delta pH$ . N, number of independent experiments on different blood pools.

the corresponding change in blood  $O_2$  content. In Fig. 5, the change in blood  $O_2$  content is equal to the  $O_2$ -carrying capacity of the oxygenated blood (8.5 vols % or 3.8 mmol  $O_2 l^{-1}$  of blood). The ratio (i.e. Haldane effect) amounts, therefore, to 0.37 mol  $CO_2 \text{ mol}^{-1} O_2$ .

Compared with voluntarily diving animals, the non-bicarbonate buffering capacity of true plasma from forcibly submerged animals increased, due to the haemoconcentration effect (Table 1). Analyses of the individual contributing slopes showed that the true plasma buffering capacity ( $\Delta[HCO_3^-]/\Delta pH$ ) is linearly related to the blood O<sub>2</sub>-carrying capacity ( $C_bO_2$ ) by the equation  $\Delta[HCO_3^-]/\Delta pH =$  $1.46C_bO_2+1.02$  (r = 0.90).

### Blood respiratory properties before, during and following enforced dives

The data in Fig. 6 summarize the time course and nature of the changes in blood characteristics following an enforced 30-min dive. Hydrogen ions from metabolic sources were clearly present after a dive because pH fell at a constant  $P_{CO_2}$  of 14.5 Torr. True plasma bicarbonate concentrations of blood tonometered at 14.5 Torr  $P_{CO_2}$  also declined during the enforced dive (Fig. 6). Using a  $HCO_3^-$ -pH diagram and the relationship between haematocrit and plasma buffering capacity, each of these post-dive bicarbonate concentrations was backtitrated along its respective buffer slope to the level which would be present at the pre-dive control pH (cf. Wood, McMahon & McDonald, 1977; McDonald, Boutilier & Toews, 1980).

The difference in bicarbonate concentration between the control pH and that after backtitration is a measure of the concentration change in metabolically originating hydrogen ions (Woodbury, 1974). Fig. 7 is a plot of the changes in true plasma concentrations of metabolic acid (in mmol 1<sup>-1</sup>) and lactate following enforced dives of the animals in Fig. 6. These measurements were made on blood samples prepared at 10% O<sub>2</sub> saturation (for time zero) and 90% O<sub>2</sub> saturation (for times B, +1, +2 and +4 h) which are realistic estimates of the state of oxygenation of the blood at these times in the experimental cycle. The Haldane effect of blood O<sub>2</sub> saturation on CO<sub>2</sub> carriage (Fig. 5) is thus taken into account in the plot of metabolic acid equivalents in Fig. 7. Changes in true plasma concentrations of metabolic acid and lactate were similar in direction but quantitatively quite different over the first 2 h following an enforced dive (Fig. 7). During that time, the lactate concentration was approximately 50% that of the metabolic acid, both subsequently declining to pre-dive levels by the fourth hour of recovery.

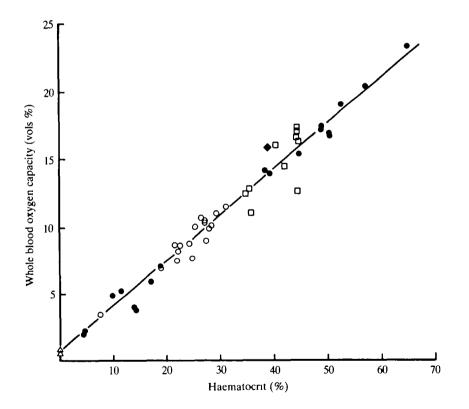


Fig. 3. Relationship *in vitro* between whole blood O<sub>2</sub>-carrying capacity and haematocrit of 2% CO<sub>2</sub>/98% air-equilibrated whole blood and separated plasma. Data shown for blood obtained from freely diving (O) and forcibly submerged ( $\Box$ ) Xenopus at 25°C. 'Haematocrit preparations' ( $\bullet$ ) represent blood whose haematocrit was experimentally altered to high and low levels by centrifugation and resuspension at various red blood cell to plasma volume ratios. Values at zero haematocrit are measurements of the O<sub>2</sub>-carrying capacity of separated plasma ( $\triangle$ ) determined at ~155 Torr P<sub>O2</sub>. See text for regression analysis.

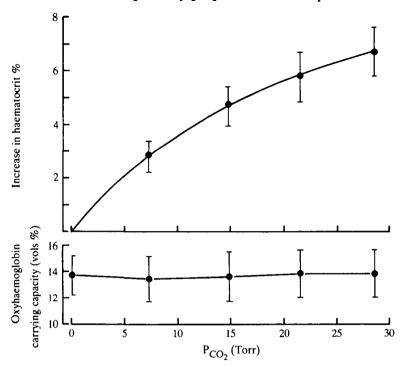


Fig. 4. Effects of step changes in equilibration gas  $P_{CO_2}$  on haematocrit of tonometered blood of *Xenopus* at 25 °C (N = 5 experiments). Haematocrit data (±1 S.E.M.) are plotted as the increase in haematocrit percentage (i.e. the difference between the haematocrit at any given level of  $P_{CO_2}$  and that of the same blood sample tonometered against air); line fitted by inspection. Lower panel shows mean (±1 S.E.M.) oxyhaemoglobin carrying capacity of same blood tonometered against each of the CO<sub>2</sub>-balanced air gas mixtures.

#### DISCUSSION

The blood O<sub>2</sub>-carrying capacity of voluntarily diving Xenopus is similar to that observed in other amphibians (Lenfant & Johansen, 1967; Wood, Weber, Maloiy & Johansen, 1975). The comparatively higher O<sub>2</sub>-carrying capacity for Xenopus blood, as reported by Jokumsen & Weber (1980), is probably explicable in terms of acute blood sampling methods since both the blood pH and haematocrit were evidently affected in the same fashion as the forcibly submerged animals of the present study. Oxyhaemoglobin affinities (at 25°C) of the semi-terrestrial anurans Rana catesbeiana ( $P_{50} = 42$  Torr at pH 7.79), Rana brevipoda ( $P_{50} = 52$  Torr at pH 7.72; Tazawa, Mochizuki & Piiper, 1979) and Bufo marinus ( $P_{50} = 53$  Torr at pH 7.75; Boutilier & Toews, 1981) are very much lower than those found at comparable temperatures and pH levels in blood from voluntarily diving Xenopus (Table 1; Fig. 1). The P<sub>50</sub> value of 29.6 Torr at pH 7.732 (Fig. 1) is similar, however, to that found in the aquatic urodele, Amphiuma tridactylum ( $P_{50} = 27$  Torr at 22°C; Lenfant & Johansen, 1967), an animal which, like Xenopus, punctuates long periods of diving with comparatively short outbursts of ventilation (Toews et al. 1971). Furthermore, the Bohr coefficient of blood from Xenopus (-0.37; Table 1; Fig. 2)is much higher than those seen in the ranids and Bufo (-0.18 to -0.27; Tazawa

et al. 1979; Boutilier & Toews, 1981). The Haldane coefficient in Xenopus blood (0.37) is identical to that of the Bohr effect, this being the prediction of the linkage equation (Wyman, 1964; Heck, 1970). Though comparable data on other anurans are not available, the linkage between coefficients suggests that a higher Bohr effect in Xenopus, relative to other more terrestrial anurans, will mean that the Haldane effect in Xenopus is also higher than in the other species. We could find no evidence for a Root effect in Xenopus whole blood, despite its apparent presence in haemo-globin solutions (Perutz & Brunori, 1982). All other interrelationships between haematocrit and O<sub>2</sub> and CO<sub>2</sub> carriage resemble conventional mammalian patterns.

The *in vitro*  $O_2$  and  $CO_2$  combining properties of blood samples taken during periods of lung ventilation were identical with those taken at various stages of voluntary dives lasting 10-30 min. In contrast, enforced 30-min dives led to a large haemoconcentration and blood acidosis (Table 1; Fig. 6). The rise in blood lactate concentrations in the forcibly submerged animal (Fig. 7) indicated that the diving  $O_2$ 

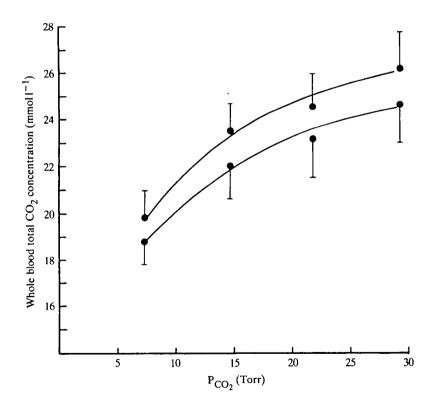


Fig. 5. In vitro relationship between the total CO<sub>2</sub> concentration of Xenopus whole blood at 25 °C as a function of the partial pressure of CO<sub>2</sub> ( $P_{CO_2}$ ). The upper curve corresponds to completely deoxygenated blood, the lower curve to blood equilibrated at  $\approx 155$  Torr  $P_{O_2}$ . At any given level of  $P_{CO_2}$  the mean total CO<sub>2</sub> concentrations (N = 5 independent experiments on separate blood pools) between oxygenated and deoxygenated blood are significantly different (Student's paired *t*-test, P < 0.05). Curved lines fitted by inspection. Values shown are means ( $\pm 1$  s.E.M.). Variations about means are due to differences in total CO<sub>2</sub> concentration between blood pools and not to variation between oxygenated and deoxygenated condition.

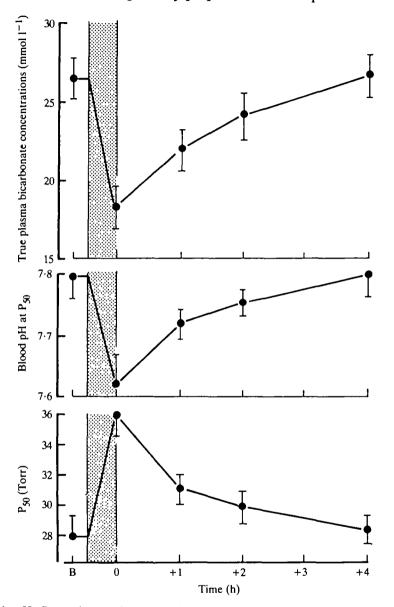


Fig. 6. pH,  $P_{50}$  and true plasma bicarbonate concentrations of blood samples taken before (B), immediately following (time 0) and +1, +2 and +4h after an enforced 30-min dive (shaded block). Blood at each stage was a pooled sample from three animals run in parallel. Four such experiments contribute to the mean (±1 s.E.M.) values shown. Blood pools at each stage were divided between oxygenated and deoxygenated tonometers whose equilibration gas mixtures were held constant at 2% CO<sub>2</sub> ( $\approx$ 14.5 Torr P<sub>CO<sub>2</sub></sub>). pH values shown correspond to P<sub>50</sub> samples prepared by the mixing method. True plasma bicarbonate concentrations were determined on blood samples prepared at P<sub>90</sub> for times B, +1, +2 and +4 at P<sub>10</sub> for time 0 (see text for further explanation).

store was not adequate to supply the full energetic demands of the dive and that a switch to anaerobiosis had occurred.

Significant differences between the true plasma concentration changes in metabolic acid and lactate following an enforced dive (Fig. 7) are reminiscent of the postactivity changes seen in the blood of other species of amphibians (McDonald *et al.* 1980; Boutilier *et al.* 1980) and some fishes (Wood *et al.* 1977; Turner, Wood & Hōbe, 1983). In all instances, the concentration of metabolic acid equivalents was significantly higher than lactate for some hours following the enforced behaviour. The most obvious resolution of this discrepancy would be if hydrogen ion production occurred *via* metabolic pathways whose end-products were other than lactate, though evidence for this is scarce (cf. Bennett, 1978). The generation of protons by metabolic processes and their degree of stoichiometry with fermentable substrates is complex and controversial (Gevers, 1977; Hochachka & Mommsen, 1983; Pörtner, Heisler & Grieshaber, 1984). The most favoured possibility for the discrepancies observed in this and other studies is that protons and lactate anions are released to and/or removed from the blood at different rates.

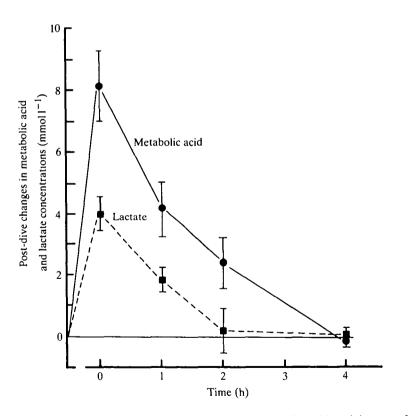


Fig. 7. Changes in true plasma concentrations of metabolic acid and lactate of the animals in Fig. 5. Note that metabolic acid (in mmoll<sup>-1</sup>) is measured as the change in true plasma bicarbonate concentration at constant  $P_{CO_2}$  plus the portion of the metabolic acid concentration which is buffered by non-bicarbonate buffers according to the *in vitro* plasma buffer slope (see Fig. 6 and text).

Experiments on working gastrocnemius muscles of Xenopus have shown in vivo and in vitro that the haemoconcentration of an enforced dive may be explained in part by plasma water uptake into the hypoxic cells of lactate-enriched muscle tissue (Boutilier, Emilio & Shelton, 1986). Presumably, a small component of the raised haematocrit in vivo results from CO2-induced red cell swelling (Fig. 4), since arterial P<sub>CO</sub>, levels are elevated by approximately 10 Torr during a 30-min enforced dive (Emilio & Shelton, 1980). There is much recent evidence that red blood cell swelling also occurs as a result of hormone-sensitive ion transport systems in erythrocytic membranes and that the associated changes in pHi, Donnan ratio, and dilution of organic polyanions can influence the respiratory properties of the blood (Palfrey & Greengard, 1981; Weber, 1982; Nikinmaa, 1983; Baroin, Garcia-Romeu, Lamare & Motais, 1984). Certainly, red cell swelling cannot account for the changes in haematocrit which lead to an elevated O2-carrying capacity of the blood and an increased non-bicarbonate buffering ability. Thus, the rise in haematocrit observed here must be largely due to haemoconcentration, increasing the buffer slope during an enforced dive (Table 1; Fig. 6) and reducing the change in blood pH that would otherwise occur. In addition, the delivery of oxygen to hypoxic tissues will be facilitated by the decreased oxyhaemoglobin affinity (Fig. 1).

Amphibians are noted for their ability to tolerate wide internal fluctuations in body oxygenation, acid-base balance and anaerobic end-product concentrations. In diving forms, these capabilities have often been viewed as metabolic adaptations that allow diving times to be greatly extended as a part of the animals' natural behaviour. Though a shift to anaerobic pathways may at times become necessary as a short-term solution to increased energy demands, this is probably avoided during most voluntary diving conditions (Shelton & Boutilier, 1982; Boutilier, 1984).

The respiratory properties of blood taken from freely moving animals seem well matched to their predominantly aquatic existence and periodic nature of air breathing. For example, the high oxyhaemoglobin affinity and Bohr and Haldane effects (Figs 1, 2, 5), relative to more terrestrial anurans (cf. Wood et al. 1975; Boutilier & Toews, 1981), can be viewed as adaptations aimed at reducing the amount of time required for effective blood gas turnover whilst ventilating at the surface. In Xenopus, these biochemical adaptations appear in unison with a cardiovascular arrangement which permits high blood flow to the lung during breathing (Shelton, 1970, 1976; Emilio & Shelton, 1972; Shelton & Boutilier, 1982). This is consistent with the observation that all movements of the buccal pump in Xenopus are concerned with lung filling and emptying whereas much of the ventilatory pattern of ranid and bufonid species involves non-pulmonary buccal oscillations (DeJongh & Gans, 1969; Macintyre & Toews, 1976; West & Jones, 1975); Xenopus precedes each inspiration phase with a single expiration, thereby ensuring a more complete renewal of the lung gas store (Brett & Shelton, 1979). Taken together, these factors seem most appropriate to an exclusively aquatic animal whose acquisition of fresh air may become limited to rather brief episodes at the water surface (Boutilier, 1984).

One obvious outcome of intermittent ventilation in *Xenopus* is that it leads to considerable oscillations in the respiratory gas tensions in blood (Emilio & Shelton,

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1974; Shelton & Boutilier, 1982). During a dive, however, the changes in P<sub>CO</sub>, are far less than those of P<sub>0</sub>, owing to the differences in the capacitances of the respiratory gases in water surrounding the permeable skin. As a result, the animal is able to operate over a wide range of blood Po,, and thus blood O2 contents, without becoming severely acidotic (Emilio & Shelton, 1974; Shelton & Boutilier, 1982; Boutilier, 1984). From the oxygen equilibrium curves in Fig. 1, it is clear that the blood O2 store will change in a complex way throughout the period of diving, as it depends not only on the oxygen utilization by the animal but also on the magnitude of the changes in P<sub>CO</sub>, and thus pH (Fig. 2). In addition, the lungs of Xenopus are used as an additional storage site for oxygen during a dive and may be selectively perfused so as to supplement an ever decreasing store of blood oxygen (Shelton & Boutilier, 1982). Just as the deoxygenated blood of *Xenopus* may be used as an effective storage site for bulk delivery of carbon dioxide to the lungs during breathing, the comparatively high Bohr effect (Table 1) represents an adaptation which will facilitate the delivery of the blood  $O_2$  store to the tissues during a dive. It seems likely that these biochemical adaptations are evolutionarily linked to the changes in respiratory blood flow during intermittent breathing which are made possible by the undivided ventricle. Thus, the Haldane effect occurs in unison with high blood flow to the lungs (Shelton, 1970), whereas the Bohr effect appears to be linked to an increasing movement of blood into the systemic circulation as apnoea progresses (Shelton & Boutilier, 1982).

Financial support from the Commonwealth Scholarship Commission is gratefully acknowledged.

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